Abiesatrines A–J: anti-inflammatory and antitumor triterpenoids from *Abies georgei* Orr†

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A novel *spiro*-lanostane (abiesatrine A, 1) was isolated from the aerial parts of *Abies georgei* together with 9 new (abiesatrines B–J, **2–10**) and 10 known triterpenes (**11–20**). The new structures were established by the extensive analysis of their spectroscopic data. The configuration of **1**, featuring a unique spirolactone formed by C-13 and C-23 *via* oxygen-bridge, was confirmed by X-ray crystallography, and its biopathway was tentatively proposed. Among these isolates, compound **16** showed the strongest inhibitory activity against LPS-induced NO production in RAW264.7 macrophages (IC₅₀ = 8.9 μ g mL⁻¹). While compounds **1** and **20** exhibited potent anti-proliferative effects on QGY-7703 cells with IC₅₀ values of 9.3 and 7.6 μ g mL⁻¹, respectively. Preliminary structure–activity relationship (SAR) investigations defined structural feature of the 24*Z*-olefinic bond key to the lanostane and cycloartane pharmacophore.

Introduction

Abies georgei occurs exclusively in the northwest of Yunnan Province and the southwest of Sichuan Province, China.¹ The CHCl₃ fraction of its EtOH extract showed significant antitumor effect, while the EtOAc part exhibited potent anti-inflammatory activity.² In the previous study on this plant, the isolation, structure, and bioactivities of flavanols, diterpenes, and norditerpenes were reported.³⁻⁵ To continuously explore the structurally and biologically novel chemical constituents from *A. georgei*, an intensive investigation was carried out, which led to the isolation of ten new (1–10) (Fig. 1) and 10 known (11–20) triterpenes. Herein, we describe the isolation, structural elucidation, and anti-inflammatory, as well as antitumor activities of these triterpenes. In addition, preliminary structure–activity relationship (SAR) for lanostanes and cycloartanes on anti-proliferative activity against QGY-7703 tumor cells were investigated.

Results and discussion

The CHCl₃- and EtOAc-soluble extracts of *Abies georgei* were subjected to column chromatography (CC) on silica gel, ODS, and Sephadex LH-20, as well as preparative TLC to afford ten

new (1–10) and ten known (11–20) triterpenes. By comparison of the ¹H and ¹³C NMR, and MS data with the published data, the known compounds were identified as (24*R*)-cycloartane-3 β ,24,25-triol (11),^{6,7} (24*R*)-cycloartane-3 α ,24,25-triol (12),⁷ methyl (24*Z*)-26-carboxy-3,4-seco-cycloarta-4(29),24-dien-3-oate (13),⁸ 23-oxomariesiic acid B (14),⁹ isofirmanoic acid (15),^{9,10} (9 β ,24*Z*)-3,23-dioxolanosta-7,24-dien-26-oic acid (16), firmanoic acid (17),⁹ 2 α ,3 β ,24- trihydroxy-12-ursen-28-oic acid (18),¹¹ dammarolic acid (19),¹² and ursolic acid (20).¹³

The molecular formula $C_{30}H_{46}O_4$ of compound 1 was established on the basis of its positive HRESIMS at m/z 493.3293 [M+Na]⁺, indicating eight degrees of unsaturation. The IR spectrum showed absorption bands characteristic of hydroxy (3384 cm⁻¹), carboxyl (1773 cm⁻¹) and olefinic bonds (1646 cm⁻¹). The ¹H and ¹³C NMR spectroscopic data of 1 (Tables 1 and 2) indicated thirty carbon signals including five singlet and two doublet methyls [$\delta_{\rm H}$ 0.89 (3H, d, J = 6.0 Hz, Me-21), 0.90 (3H, s, Me-19), 0.91 (3H, s, Me-29), 0.95 (3H, s, Me-28), 0.98 (3H, s, Me-18), 1.13 (3H, s, Me-30), 1.32 $(3H, d, J = 7.2 \text{ Hz}, \text{Me-26}); \delta_{C} 15.7 (q, \text{Me-21}), 16.0 (q, \text{Me-26}),$ 17.3 (q, Me-18), 21.7 (q, Me-30), 21.8 (q, Me-19), 22.5 (q, Me-29), 27.8 (q, Me-28)], nine methylenes, six methines (one olefinic carbon and one oxygenated sp³ methines), and eight quaternary carbons including one carboxyl ($\delta_{\rm C}$ 179.0), one sp² carbon ($\delta_{\rm C}$ 143.0), and two oxygenated sp³ quaternary carbons ($\delta_{\rm C}$ 90.9 and 109.0). In the ¹H–¹H COSY spectrum, six fragments were obtained according to the spin systems of H₂-1/H₂-2/H-3, H-5/H₂-6/H-7, H-9/H₂-11/H₂-12, H₂-15/H₂-16, H₃-21/H-20/H₂-22, and H₂-24/H-25/H₃-26 (Fig. 2). According to the correlations traced from seven methyls (Me-18,19,21,26,28,29,30) and the olefinic proton ($\delta_{\rm H}$ 5.48, dd, J = 5.4, 2.4 Hz, H-7), the structure of 13,17-friedo-lanostane triterpenoid was established for compound 1. According to the unsaturation degrees ($\Omega = 8$) and signals for a hemiketal moiety ($\delta_{\rm C}$ 109.0, C-23) together with an oxygenerated quaternary carbon (δ_c 90.9, C-13), the planar structure of 1 was supposed to contain a novel spirolactone moiety. Further

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Fig. 1 Structures of compounds 1–10.

evidence for its relative configuration was confirmed undoubtedly by the X-ray diffraction analysis[‡] (Fig. 3). Therefore, **1** was elucidated as 13,17-*friedo*-13,23-epoxy-3 α -hydroxy-9 β -lanosta-7ene-23,27-olide, named abiesatrine A. It is the first example of lanostane triterpenoid bearing a unique spirolactone formed by C-13 and C-23 *via* oxygen-bridge. A tentative biosynthetic pathway is proposed in Scheme 1. It might be synthesized from **2** *via* hydration reaction and aldol condensation to give 13,17-*friedo*-13,23-epoxy-3 α ,23-dihydroxy-9 β -lanosta-7,24-diene-27-oic acid. Then intramolecular lactonization of the intermediate followed by reduction constructed the spirolactone system of abiesatrine A.



Fig. 2 Key ¹H-¹H COSY, HMBC, and NOESY correlations for 1.



Fig. 3 X-Ray structure of abiesatrine A (1). Displacement ellipsoids are drawn at the 50% probability level.

Compound 2 exhibited the molecular formula $C_{30}H_{44}O_4$ as evidenced by the positive HRESIMS at m/z 491.3123 [M+Na]⁺. It showed ¹H and ¹³C NMR spectra similar to those of 14. However, close comparison of the ¹³C NMR spectroscopic data between 2 and 14 revealed significant differences: C-24 was shifted downfield by 2.6 ppm, while C-25 upfield 1.5 ppm. Such phenomena were also found in other E/Z-configuration pairs of the lanostanes.⁹ Thus, the configuration of the olefinic bond at C-24 in 2 should be Z compared to E in 14. In the NOESY spectrum, H-24 was correlated to H₃-27 (Fig. 4), which confirmed the presence of Z-orientated olefinic bonds at C-24 in 2. On the basis of above evidences, compound 2 was then established as 13,17-*friedo*-3 α -hydroxy-9 β -lanosta-7,12,24Ztriene-23-oxo-27-oic acid, named abiesatrine B.

Compound **3** shared the same molecular formula as **2**. And they exhibited similar IR and NMR data. However, differences were found by careful comparison of their ¹H and ¹³C NMR spectroscopic data: the methyl at C-26 and vinyl methine at C-24 in **2** were altered to be one olefinic [$\delta_{\rm H}$ 6.04 (1H, d, J = 2.1 Hz), 5.37 (1H, d, J = 2.1 Hz); $\delta_{\rm C}$ 124.4 t] and one aliphatic [$\delta_{\rm H}$ 3.51 (2H,

[‡] X-Ray diffraction analysis of abiesatrine A (1): colorless orthorhombic crystal of C₃₀H₄₆O₄·CH₃OH. Space group *P*1, *a* = 8.165(15) Å, *b* = 13.02(3) Å, *c* = 26.83(6) Å, *V* = 2852(10) Å³, *Z* = 4; crystal size 0.15 × 0.10 × 0.04 mm³. A total of 11872 unique reflections (θ = 1.52–25.01°) were collected using graphite monochromated Mo-Kα (λ = 0.71073 Å) on a CCD area detector diffractometer. The structure was solved by direct methods (SHELXS-97) and expanded using Fourier techniques (SHELXS-97). The final cycle of full-matrix least-squares refinement was based on 5047 data, 0 restraints and 327 variable parameters. Final *R* indicates *R*₁ = 0.0711, w*R*₂ = 0.1422 [*I* > 2σ(*I*)]. Crystallographic data (excluding structure factors) for the structure of abiesatrine A (1) in this paper are available as ESI.[†]

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Table 1 ¹H NMR spectroscopic data for compounds 1-10 (J in Hz within parentheses)

			,							
No.	1ª	2^{b}	36	4ª	5 <i>a</i>	6 ^c	7 ^b	8 ^c	96	10 ⁶
-1	1.97 m; 0.84 m	2.28 m; 1.83 m	1.28 brs; 0.95 m	1.73 m; 1.61 m	1.80 m; 1.41 m	1.88 m; 1.43 m	2.52 m; 1.60 m	2.52 m; 1.75 m	2.13 m; 1.90 m	2.48 m; 2.00 m
0	1.60 m; 1.35 m	1.95 m; 1.58 m	2.01 m; 1.57 m	2.50 (dt, 7.5, 1.8)	1.64 m	1.66 m; 1.59 m	2.44 m; 1.57 m	2.46 m; 1.63 m	1.61 m; 1.57 m	2.43 m; 2.18 m
б	3.44 (t, 3.0)	3.39 m	3.38 m		3.22 (dd, 10.2, 5.4)	3.12 (dd, 11.4, 4.5)			2.87 (t, 3.0)	
S	1.73 m	$1.45 \mathrm{m}$	1.42 m	1.42 (dt, 12.0, 1.2)	0.85 (dd, 10.8, 3.6)	0.87 (dd, 12.0, 4.2)	$1.45 \mathrm{m}$	1.46 m	1.81 (dd, 12.0, 4.8)	2.51 (dd, 12.0, 5.4)
9	$1.94 \mathrm{m}$	$1.92 \mathrm{m}$	1.92 m	1.94 m; 1.89 m	$1.94 \mathrm{m}$	1.92 m	1.92 m; 1.79 m	1.93 m; 1.79 m	1.54 m; 0.79 m	1.48 m
7	5.48 (dd, 5.4, 2.4)	5.66 m	5.66 m	5.65 (dt, 7.8, 2.7)	5.57 (dt, 7.2, 3.0)	5.58 (dt, 7.2, 1.8)	5.66 (dt, 8.1, 2.8)	5.67 (dt, 7.8, 3.0)	1.58 m; 1.03 m	2.08 m; 1.30 m
8									1.59 m	1.59 m
6	1.60 m	1.42 m	1.45 m	2.21 m	2.27 m	2.29 (brd, 13.2)	2.24 m	2.25 m		
11	1.71 m	2.25 m	2.34 m	1.64 m	1.91 m	1.75 m	1.66 m	1.66 m	2.03 m	2.23 m
	$1.35 \mathrm{m}$	1.81 (dd, 14.7, 2.4)	0 1.82 m		1.69 m	1.45 m			1.06 m	1.33 m
12	1.95 m; 1.79 m	5.56 (dd, 8.4, 2.4)	5.55 (dd, 8.1, 2.4)	1.85 m; 1.72 m	1.56 m; 1.16 m	1.82 m; 1.68 m	1.64 m	1.87 m; 1.66 m	1.63 m	1.68 m; 1.58 m
15	1.93 m; 1.15 m	1.45 m	1.53 m; 1.42 m	1.60 m; 1.43 m	1.60 m; 1.51 m	1.49 m; 1.23 m	1.55 m; 1.42 m	1.58 m; 1.44 m	1.31 m	1.32 m
16	1.92 m	1.92 m	1.91 m	1.96 m	1.95 m	2.00 m	2.05 m	2.04 m	1.31 m	1.32 m
	$1.54 \mathrm{m}$	$1.44 \mathrm{m}$	1.43 m	1.29 m	1.26 m	1.52 m	1.28 m	1.26 m	1.09 m	
17				1.54 m	1.46 m	1.52 m	1.56 m	1.58 m	1.63 m	1.64 m
18	0.98 s	0.96 s	0.94 s	0.79 s	0.91 s	0.89 s	0.77 s	0.77 s	1.00 s	0.99 s
19	0.90 s	0.95 s	0.95 s	1.00 s	0.98 s	0.99 s	0.98 s	0.99 s	0.52 (d, 4.2)	0.73 (d, 4.2)
									0.34 (d, 4.2)	0.41 (d, 4.2)
20	1.80 m	2.21 m	2.20 m	1.42 m	1.42 m	1.33 m	1.35 m	1.31 m	1.38 m	1.43 m
21	0.89 (d, 6.0)	0.88 (d, 6.3)	0.87 (d, 6.3)	0.92 (d, 6.6)	0.91 (d.6.6)	0.95 (d, 6.6)	0.96 d (d, 6.3)	0.97 (d, 6.0)	0.91 (d, 6.0)	0.94 (d, 6.0)
22	1.77 m	2.92 (dd, 14.1, 1.8)	1 2.86 (dd, 16.0, 2.7)	1.63 m	1.43 m	1.63 m	1.63 m	1.64 m	1.01 m	1.32 m
	1.73 (d, 10.2)	2.23 m	2.34 (d, 16.0)	1.58 m	1.30 m	1.39 m	1.38 m	1.41 m		1.18 m
23				2.24 m; 2.15 m	2.25 m; 2.11 m	4.51 (dt, 9.6, 4.5)	4.50 (dt, 9.3, 4.5)	4.52 (dt, 9.6, 4.8)	1.90 m; 1.31 m	2.20 m; 2.08 m
24	2.29 (dd, 12.6, 9.0 2.00 (dd, 12.6, 7.2)) 6.86 (d, 1.5)	3.51 s	6.91 (dt, 7.5, 1.2)	6.90 (t, 7.2)	6.55 (dd, 9.6, 1.2)	6.45 (dd, 9.3, 1.2)	6.56 (dd, 9.0, 1.2)	3.16 (dd, 10.2, 1.8)	6.64 (t, 7.8)
25	2.66 m									
26	1.32 (d, 7.2)	2.16 (d, 1.5)						1.86 (d, 1.2)	1.12 s	
27			6.04 (d, 2.1) 5.37 (d, 2.1)	1.85 br s	1.85 s	1.86 (d, 1.2)	1.86 (d, 1.2)		1.15 s	1.80 s
28	0.95 s	0.94 s	0.94 s	1.10 s	1.02 s	0.98 s	$1.07 \mathrm{s}$	1.08 s	0.91 s	1.69 s
29	0.91 s	0.92 s	0.93 s	1.11 s	0.88 s	0.85 s	1.08 s	1.09 s	0.89 s	4.84 (d, 2.4)
30 OM	1.13 s e	1.20 s	1.18 s	1.03 s	1.03 s	1.05 q	1.05 s	1.06 s	0.94 s 3.31 s	1.01 s
" Me	asured at 600 MHz	in CDCl ₃ . ^b Measure	ed at 300 MHz in CD)3OD. ^e Measured a	tt 600 MHz in CD ₃ C	Q				

 Table 2
 ¹³C NMR spectroscopic data for compounds 1–10

No.	1 <i>ª</i>	2 ^{<i>b</i>}	3 ^b	4 ^{<i>a</i>}	5 ^{<i>a</i>}	6 ^c	7 ^b	8 ^c	9 ^b	10 ^b
1	28.1 t	30.7 t	30.7 t	34.2 t	35.2 t	36.9 t	35.2 t	35.2 t	28.7 t	31.2 t
2	25.2 t	26.5 t	26.5 t	34.3 t	27.9 t	28.6 t	35.3 t	35.3 t	29.1 t	34.4 t
3	76.6 d	77.2 d	77.2 d	219.1 s	79.4 d	80.0 d	221.5 s	221.7 s	88.0 d	180.4 s
4	36.9 s	38.0 s	38.0 s	47.0 s	38.8 s	39.9 s	48.1 s	48.1 s	40.9 s	150.9 s
5	36.5 d	39.4 d	39.4 d	52.4 d	48.6 d	50.0 d	53.7 d	53.7 d	42.9 d	47.0 d
6	23.8 t	24.3 t	24.3 t	23.0 t	23.1 t	24.2 t	23.9 t	23.9 t	22.0 t	29.1 t
7	118.4 d	120.1 d	120.0 d	121.5 d	121.5 d	122.9 d	122.7 d	122.8 d	29.2 t	26.2 t
8	143.0 s	147.5 s	147.6 s	148.7 s	148.8 s	150.1 s	149.9 s	150.0 s	49.4 d	49.6 d
9	48.3 d	52.6 d	52.7 d	45.5 d	48.4 d	49.8 d	46.8 d	46.8 d	21.0 s	22.6 s
10	34.8 s	36.0 s	36.0 s	35.8 s	35.9 s	37.0 s	36.9 s	37.0 s	27.8 s	28.6 s
11	25.9 t	29.1 t	29.1 t	20.9 t	22.9 t	24.0 t	21.9 t	21.9 t	26.8 t	28.1 t
12	34.7 t	123.8 d	123.6 d	34.4 t	34.6 t	36.6 t	35.6 t	35.6 t	34.2 t	34.4 t
13	90.9 s	157.4 s	157.4 s	44.0 s	43.6 s	44.8 s	45.1 s	45.2 s	46.4 s	46.4 s
14	53.9 s	51.2 s	51.1 s	51.9 s	52.7 s	54.0 s	53.1 s	53.1 s	50.1 s	50.2 s
15	33.7 t	37.9 t	37.9 t	33.1 t	33.4 t	34.5 t	34.2 t	34.2 t	36.6 t	36.8 t
16	39.3 t	39.2 t	39.1 t	29.7 t	28.6 t	30.0 t	29.6 t	29.6 t	27.4 t	29.2 t
17	45.1 s	47.6 s	47.5 s	53.0 d	53.4 d	55.4 d	55.0 d	54.9 d	53.7 d	53.6 d
18	17.3 q	25.4 q	25.6 q	22.4 q	23.6 q	24.1 q	22.9 q	22.8 q	18.5 q	18.7 q
19	21.8 q	22.8 q	22.8 q	23.1 q	24.5 q	24.9 q	23.5 q	23.5 q	30.6 t	31.0 t
20	34.6 đ	40.2 đ	39.4 đ	36.1 đ	36.1 đ	34.8 đ	34.7 đ	34.8 đ	37.8 d	37.2 d
21	15.7 q	16.2 g	16.2 q	18.2 q	18.2 q	20.0 q	20.0 g	20.0 g	19.0 q	18.7 q
22	37.8 t	49.4 t	46.9 t	34.6 t	35.5 t	44.7 t	44.8 t	44.7 t	34.9 t	36.3 t
23	109.0 s	205.5 s	212.2 s	26.0 t	26.0 t	67.5 d	67.6 d	67.5 d	30.7 t	26.5 t
24	44.8 t	129.7 d	48.2 t	145.7 d	145.6 d	144.8 d	142.6 d	145.0 d	80.6 d	141.7 d
25	35.3 d	149.1 s	141.8 s	126.6 s	126.6 s	129.9 s	132.1 s	129.6 s	73.9 s	130.7 s
26	16.0 q	15.9 q	173.9 s	172.6 s	172.7 s	171.7 s	173.6 s	13.2 q	25.0 q	174.1 s
27	179.0 s	174.6 s	124.4 t	12.0 q	12.0 q	13.2 q	13.7 q	171.4 s	25.7 q	12.9 q
28	27.8 q	28.9 q	28.9 q	28.0 q	28.9 q	29.5 q	28.4 q	28.4 q	26.4 q	20.2 q
29	22.5 q	23.6 q	23.6 q	21.3 q	16.4 q	17.0 q	21.7 q	22.0 q	22.0 q	112.0 t
30	21.7 q	26.6 q	26.6 q	27.4 q	30.5 q	31.0 q	27.9 q	19.8 q	19.8 q	19.8 q
OMe	<u>^</u>	-	•	•	•	-	-	•	57.4 a	

^a Measured at 150 MHz in CDCl₃. ^b Measured at 75 MHz in CD₃OD. ^c Measured at 150 MHz in CD₃OD



Scheme 1 Postulated biosynthetic pathway for abiesatrine A (1).



Fig. 4 Selected ${}^{1}H{-}^{1}H$ COSY, HMBC, and NOESY correlations for 2.

s); $\delta_{\rm C}$ 48.2 t] methylenes in **3**. This suggested that the vinyl bond in **3** should be located at positions of C-25 and C-27, instead of C-24,25 in **2**. By detailed analysis of the HMBC NMR spectrum

(Fig. 5), compound **3** was then concluded to be 13,17-friedo-3 α -hydroxy-9 β -lanosta-7,12,25(27)-trien-23-oxo-26-oic acid, named abiesatrine C.



Fig. 5 Key HMBC correlations for compound **3**.

Compound **4** was found to possess the molecular formula $C_{30}H_{46}O_3$, as evidenced by its negative HRESIMS at m/z 453.3320 $[M - H]^-$. The ¹H and ¹³C NMR spectra showed high similarity to those of (24*E*)-3-oxo-9 β -lanosta-7,24-dien-26-ol except for a carboxy group (δ 172.6 s) at C-26 of **4** instead of an oxymethylene moiety in (24*E*)-3-oxo-9 β -lanosta-7,24-dien-26-ol.¹⁴ This assumption was confirmed by the HMBC correlations of H₃-27 to the carboxyl (Fig. 6). Therefore, compound **4** was assigned as (24*E*)-3-oxo-9 β -lanosta-7,24-dien-26-ol assigned assigned assigned as (24*E*)-3-oxo-9 β -lanosta-7,24-dien-26-ol assigned assigned assigned assigned as (24*E*)-3-oxo-9 β -lanosta-7,24-dien-26-ol assigned ass



Fig. 6 Key ¹H–¹H COSY and HMBC correlations for compound 4.

Compound **5** gave the molecular formula $C_{30}H_{48}O_3$ from the negative HRESIMS at m/z 455.3507 [M – H]⁻. Its ¹H and ¹³C NMR spectroscopic data were very similar to those of **4** except that an sp² ketone group in **4** was changed to be an sp³ oxymethine [$\delta_{\rm H}$ 3.22 (1H, dd, J = 10.8, 5.4 Hz); $\delta_{\rm C}$ 79.4 d] in **5**. This indicated a hydroxy moiety at C-3 in **5** instead of a ketone in **4**. This assumption was confirmed by the HMBC correlations of H₃-28,29 to the oxymethine at $\delta_{\rm C}$ 79.4 (Fig. 7). According to the large coupling constant (dd, ${}^{3}J_{\rm H2-H3} = 10.8$, 5.4 Hz) of H-2 and H-3, the configuration of 3-OH was deduced as β -orientated. Further evidences were obtained in the NOESY spectrum by the correlations of H-3 to H-1 α ,2 α ,5, H₃-28, and H₃-19 to H₃-29, H-9 (Fig. 7). Accordingly, the structure of compound **5** was defined as (24*E*)-3 β -hydroxy-9 β -lanosta-7,24-dien-26-oic acid, named abiesatrine E.



Fig. 7 Selected HMBC and NOESY correlations for compound 5.

Compound **6** had a molecular formula $C_{30}H_{48}O_4$ as established from its negative HRESIMS at m/z 471.3453 [M – H]⁻. The ¹H and ¹³C NMR spectroscopic data were related closely to those of **5**, with the only difference being the presence of an additional hydroxy group at C-23 position [$\delta_{\rm H}$ 4.51 (1H, dt, J = 9.6, 4.5 Hz); $\delta_{\rm C}$ 67.5 d]. This can be confirmed in the ¹H–¹H COSY spectrum by the correlations of vinyl proton ($\delta_{\rm H}$ 6.55, 1H, dd, J = 9.6,1.2 Hz) to oxymethine ($\delta_{\rm H}$ 4.51, 1H, dt, J = 9.6, 4.5 Hz) and H₃-27 ($\delta_{\rm H}$ 1.86, 3H, d, J = 1.2 Hz) (Fig. 8). The *R*-configuration of C-23 was determined mainly based on the coupling constant (J = 9.6 Hz) of H-23 and H-24.¹⁵ As such, compound **6** was deduced to be (23R, 24E)-3 β , 23-dihydroxy-9 β -lanosta-7, 24-dien-26-oic acid, named abiesatrine F.



Fig. 8 Selected ${}^{1}H{-}{}^{1}H$ COSY, HMBC and NOESY correlations for compound 6.

Compound 7 presented a molecular formula of $C_{30}H_{46}O_4$ by positive HRESIMS at m/z 493.3272 [M+Na]⁺. The ¹H and ¹³C NMR spectra consisted of signals similar to those of **6** except for a ketone moiety (δ 221.5 s) at C-3 instead of a hydroxy group in **6**. By detailed analysis of its HSQC, ¹H-¹H COSY, HMBC, and NOESY spectra (Fig. 9), **7** was thus assigned as (23*R*,24*E*)-3-oxo-9 β -lanosta-7,24-dien-23-hydroxy-26-oic acid, named abiesatrine G.



Fig. 9 Key ¹H-¹H COSY, HMBC and NOESY correlations for 7.

Compound **8** gave the same molecular formula as **7**. In addition, it shared almost the same IR, ¹H and ¹³C NMR spectroscopic data as those of **7**. However, a close inspection of their ¹³C NMR spectroscopic data revealed significant differences: downshift of C-24 by 2.4 ppm, while upshift of C-25 by 2.5 ppm. This suggested the presence of a *Z*-oriented vinyl bond at C-24 of **8** instead of *E*-configuration in **7**. The similar variations because of the differences of *E*/*Z*-configuration can be found in compounds **2** and **14**, as well as the other lanostanes.⁹ In the NOESY spectrum, H₃-26 was correlated to H-24 (Fig. 10), which confirmed the presence of a 24*Z*-olefinic bond. Consequently, compound **8** was established to



Fig. 10 Selected NOESY correlations for compound 8.

be (23R,24Z) 3-oxo-9 β -lanosta-7,24-dien-23-hydroxy-27-oic acid, named abiesatrine H.

Compound 9 gave a molecular formula $C_{31}H_{34}O_3$ deduced from its positive HRESIMS at m/z 497.4057 [M+Na]⁺. It exhibited very similar IR, ¹H and ¹³C NMR spectra to those of **12**⁷ except for an additional methoxyl [δ_{H} 3.31 (3H, s); δ_{C} 57.4 (q)]. Compared to **12**, C-3 of **9** was upshifted by 10.3 ppm, which established the connection of the methoxyl to C-3 position. By its HMBC correlations (Fig. 11), compound **9** was then identified to be (24*R*)-3 α -methoxycycloartane-24,25-diol, named abiesatrine I.



Fig. 11 Key HMBC correlations for compound 9.

Compound **10** showed a molecular ion peak at m/z 469.3321 $[M - H]^-$ in its negative HRESIMS, corresponding to the molecular formula of $C_{30}H_{46}O_4$. Its ¹H and ¹³C NMR spectroscopic data were very similar to those of **13**⁸ except for the presence of a carboxy group instead of methyl ester moiety at C-3, which can be confirmed according to the downshift of C-2 and C-3 by 2.0 and 4.2 ppm, respectively. In addition, C-25 was downshifted by 2.1 ppm, while C-24 upshifted by 2.7 ppm. According to the similar phenomena in **2/14**, **7/8**, and other triterpenes,⁹ compound **10** was supposed to bear an *E*-configuration olefinic bond at C-24,25. By detailed analysis of its 2D NMR spectra (Fig. 12), **10** was then concluded to be (24*E*)-3,4-secocycloarta-4(28),24-dien-3,26-dioic acid, named abiesatrine J.



Fig. 12 Key ${}^{1}H{}^{-1}H$ COSY (bold) and HMBC (arrow) correlations for compound 10.

Lanostanes are the main triterpenes in *Abies* plants together with minor cycloartanes, gammacerane, and hopanes.^{16,17} This is coincident with the results obtained from this study on *A. georgei*. Briefly, among the isolated 20 triterpenes, 12 are lanostanes (1–8 and 14–17) and 5 cycloartanes (9–13). The other three are ursanes (18–20), which is a new type of triterpenes found in *Abies* species for the first time.

Since the CHCl₃ fraction of the EtOH extract of *A. georgei* showed potent antitumor effect against QGY-7703 and LOVO

cells, while the EtOAc part exhibited strong inhibitory activity on nitric oxide (NO) induction by lipopolysaccharide (LPS) in RAW264.7 macrophages,² all the isolates (1-20) were then subjected to these two bioassays. For inhibitory effects against LPS-induced NO production, compound 16 showed remarkably active (IC₅₀ = 8.9 μ g mL⁻¹), while 15 and 17 exhibited weak activities with IC₅₀ values of 19.8 and 17.6 µg mL⁻¹, respectively. For antitumor experiments against QGY-7703 and LOVO cells, compounds 1 and 20 displayed a significant effect with IC₅₀ values of 9.3 and 7.6 µg mL⁻¹, respectively. However, only 20 showed moderate activity on LOVO cells (IC₅₀ = 14.7 μ g mL⁻¹). It is interesting to note that compounds displaying anti-inflammatory activities were all isolated from EtOAc fraction, while those with antitumor activities were from CHCl₃ extract (except 16). This is coincident with the previous bioactive results of the crude extracts of A. georgei.²

Experimental

General

NMR spectra were recorded on a Bruker Avance 600 or Avance 300 NMR spectrometer with TMS as internal standard. ESIMS were measured on an Agilent LC/MSD Trap XCT spectrometer (Waters, USA), and HRESIMS on a Q-TOF micro mass spectrometer (Waters, USA). Optical rotations were acquired with Perkin-Elmer 341 polarimeter. IR spectra were recorded on a Bruker Vector-22 spectrometer with KBr pellets. Materials for CC were silica gel (Huiyou Silical Gel Development Co. Ltd., Yantai, China), Sephadex LH-20 (Amersham Pharmacia Biotech AB, Uppsala, Sweden), and YMC-GEL ODS-A (YMC, USA). Prep. TLC was conducted with glass precoated silica gel GF₂₅₄ (Yantai).

Plant material

The aerial parts of *A. georgei* Orr were collected from Zhongdian city, Yunnan Province of China in July 2006, and were identified by Prof. Li-Shang Xie in Kunming Institute of Botany, Chinese Academy of Sciences. A herbarium specimen (No. 2006-07-016) was deposited in School of Pharmacy, Second Military Medical University, China.

Extraction and isolation

The plant material (22 kg) was pulverized and extracted with 80% EtOH under reflux for 3×3 h. The extracts were combined to concentrate to a small volume and then partitioned with CHCl₃ (25 L), EtOAc (40 L), and n-BuOH (50 L), respectively. The EtOAc extract (282 g) was separated into six fractions (F₁–F₆) by CC over silica gel eluting with gradient CHCl₃–Me₂CO. Fraction F₁ (36.3 g) was subjected to CC over MCI, Sephadex LH-20, and silica gel to give **16** (195.2 mg) and **17** (100.4 mg). Fraction F₂ was divided into 20 subfractions (F₂₋₁–F₂₋₂₀) by RP-MPLC eluting with MeOH–H₂O (5:95–100:0). Compounds **9** (8.6 mg), **11** (85.4 mg), and **15** (44.2 mg) were obtained after CC over LH-20 (CHCl₃–MeOH, 1:1 and 0:1) followed by repeated prep. TLC with CHCl₃–MeOH (20:1) from subfractions F₂₋₃, F₂₋₁₆, and F₂₋₁₇, respectively. After CC on LH-20 with CHCl₃–MeOH (1:1) and MeOH, subfraction F₂₋₂₀ was then subjected to prep. TLC using petroleum ether

(PE)–EtOAc (1:1) and CHCl₃–MeOH (20:1) to give **10** (16.0 mg) and **12** (7.1 mg). Similarly, **3** (20.5 mg) and **7** (35.8 mg) were isolated from fraction F₃. The CHCl₃ extract (906 g) was separated into five fractions (F_{C1} – F_{C5}) by CC over silica gel eluting with gradient PE–CHCl₃. Compounds **1** (52.6 mg), **4** (16.0 mg), and **5** (9.0 mg) were purified from fraction F_{C2} after repeated CC over LH-20 eluting with CHCl₃–MeOH (11:1) and MeOH, followed by prep. TLC using CHCl₃–MeOH (100:1) and/or PE–EtOAc (4:1). By the similar procedures, **2** (90.7 mg), **6** (13.5 mg), **8** (213.0 mg), **13** (37.0 mg), and **14** (116.6 mg) were isolated from fraction F_{C4} , and **18** (28.2 mg), **19** (29.2 mg), **20** (21.8 mg) from fraction F_{C5} , respectively.

Abiesatrine A (1). Colorless orthorhombic crystals; $[\alpha]_D^{20}$ –42.1 (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 216 (4.73); IR (KBr) v_{max} 3384, 2965, 2928, 2873, 1773, 1646, 1383, 1308, 1203, 1166, 1084, 884 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 471 [M+H]⁺, 493 [M+Na]⁺, 963 [2M+Na]⁺; HRESIMS (positive) [M+Na]⁺ m/z 493.3293, calcd for C₃₀H₄₆O₄Na, 493.3294.

Abiesatrine B (2). Amorphous powder; $[\alpha]_{D}^{20} - 82.7$ (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 216 (4.08); IR (KBr) v_{max} 3433, 2965, 2924, 2869, 1626, 1446, 1370, 1063, 986, 832 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 491 [M+Na]⁺; ESIMS (negative) m/z 467 [M - H]⁻, 935 [2M - H]⁻; HRESIMS (positive) [M+Na]⁺ m/z 491.3123, calcd for C₃₀H₄₄O₄Na, 491.3137.

Abiesatrine C (3). Amorphous powder; $[\alpha]_D^{20} - 31.4$ (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 211 (3.92); IR (KBr) v_{max} 3432, 2965, 2925, 2830, 1606, 1587, 1364, 1064, 985, 832 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) *m/z* 491 [M+Na]⁺; ESIMS (negative) *m/z* 467 [M – H]⁻; HRESIMS (positive) [M+Na]⁺ *m/z* 491.3156, calcd for C₃₀H₄₄O₄Na, 491.3137.

Abiesatrine D (4). Amorphous powder; $[\alpha]_{D}^{20} + 30.5$ (*c* 0.60, MeOH); UV (MeOH) λ_{max} (log ε): 215 (4.02); IR (KBr) v_{max} 3411, 2933, 2877, 1736, 1708, 1645, 1457, 1420, 1383, 1281, 1217, 1104, 815 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 477 [M+Na]⁺; ESIMS (negative) m/z 453 [M – H]⁻; HRESIMS (negative) [M – H]⁻ m/z 453.3320, calcd for C₃₀H₄₅O₃, 453.3369.

Abiesatrine E (5). Amorphous powder; $[\alpha]_D^{0} - 9.8$ (*c* 0.32, MeOH); UV (MeOH) λ_{max} (log ε): 211 (3.91); IR (KBr) v_{max} 3447, 2969, 2944, 2870, 1738, 1685, 1653, 1456, 1365, 1228, 1217, 1026, 911 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 479 [M+Na]⁺; ESIMS (negative) m/z 455 [M - H]⁻, 911 [2M - H]⁻; HRESIMS (negative) [M - H]⁻ m/z 455.3507, calcd for C₃₀H₄₇O₃, 455.3525.

Abiesatrine F (6). Amorphous powder; $[\alpha]_{D}^{20} - 18.0$ (*c* 0.21, MeOH); UV (MeOH) λ_{max} (log ε): 212 (4.05); IR (KBr) v_{max} 3405, 2945, 2927, 2874, 2621, 1698, 1682, 1651, 1463, 1384, 1259, 1216, 1162, 1052, 964 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 495 [M+Na]⁺; ESIMS (negative) m/z 471 [M – H]⁻, 943 [2M – H]⁻; HRESIMS (negative) [M – H]⁻ m/z 471.3453, calcd for C₃₀H₄₇O₄, 471.3474.

Abiesatrine G (7). Amorphous powder; $[\alpha]_{D}^{20}$ +54.5 (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 212 (4.17); IR (KBr) v_{max}

3388, 2975, 2930, 2872, 1707, 1625, 1547, 1463, 1383, 1153, 1054, 817 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 493 [M+Na]⁺; ESIMS (negative) m/z 469 [M – H]⁻, 939 [2M – H]⁻; HRESIMS (positive) [M+Na]⁺ m/z 493.3272, calcd for C₃₀H₄₆O₄Na, 493.3294.

Abiesatrine H (8). Amorphous powder; $[\alpha]_D^{20} + 63.9$ (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 215 (4.12); IR (KBr) v_{max} 3387, 2973, 2928, 2873, 1741, 1708, 1648, 1563, 1514, 1456, 1384, 1149, 877 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 471 [M+H]⁺, 493 [M+Na]⁺, 963 [2M+Na]⁺; ESIMS (negative) m/z 469 [M – H]⁻, 939 [2M – H]⁻; HRESIMS (negative) [M – H]⁻ m/z 469.3311, calcd for C₃₀H₄₅O₄, 469.3318.

Abiesatrine I (9). Amorphous powder; $[\alpha]_{D}^{20}$ -6.6 (*c* 0.50, MeOH); UV (MeOH) λ_{max} (log ε): 211 (3.79); IR (KBr) v_{max} 3446, 2931, 2869, 1700, 1651, 1540, 1457, 1383, 1100 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 497 [M+Na]⁺, 971 [2M+Na]⁺; ESIMS (negative) m/z 509 [M+Cl]⁻; HRESIMS (positive) [M+Na]⁺ m/z 497.4057, calcd for C₃₁H₅₄O₃Na, 497.3971.

Abiesatrine J (10). Amorphous powder; $[\alpha]_D^{20} + 23.0$ (*c* 0.36, MeOH); UV (MeOH) λ_{max} (log ε): 215 (4.29); IR (KBr) v_{max} 3445, 2929, 2870, 1771, 1733, 1646, 1559, 1384, 1277, 889 cm⁻¹; for ¹H and ¹³C NMR data, see Tables 1 and 2; ESIMS (positive) m/z 493 [M+Na]⁺; ESIMS (negative) m/z 469 [M – H]⁻; HRESIMS (negative) [M – H]⁻ m/z 469.3321, calcd for C₃₀H₄₅O₄, 469.3318.

Assays for anti-inflammatory and antitumor activities

Experiments were carried out according to the previously reported procedures of inhibitory activities against LPS-induced NO production in RAW264.7 macrophages and antitumor activities on QGY-7703 and LOVO cell lines.^{2,3} The cell viability was evaluated by MTT reduction, and the absorbances at 570 nm were read using a microtiter plate reader.

Conclusions

Triterpenoids are the main chemical constituents of *Abies* species. From 19 plants of this genus, about 277 compounds were identified, among which 74 were triterpenes and 36 diterpenes.¹⁷ Interestingly, in our recent research for *Abies* plants indigenous in China, many diterpenoids (including norditerpenoids) were isolated.^{3,5,18} Since norditerpenoids are not found before from this genus, it is of important chemotaxonomic significance for the norditerpenoids as the characteristics of the *Abies* plants in China.

A comparison of the cytotoxicity of all the isolates (1–20) against QGY-7703 cancer cells were carried out. Among 12 lanostane triterpenoids, abiesatrines B (2), H (8), and (9 β ,24Z)-3,23-dioxolanosta-7,24-dien-26-oic acid (16) displayed positive antiproliferative activities. The other lanostanes (except 1), however, were negative (Table 3). This highlights the structure–activity relationship (SAR) importance of the 24Z-olefinic bond. Such SAR can also be extended to cycloartane triterpenoids (*i.e.* for five cycloartanes, only 13 showed positive effect, while compounds 9–12 were negative). Furthermore, close comparison of the cytotoxicity data for 1, 2, 6, and 8 revealed a pharmacophore bias

Table 3Antiproliferative activities of 20 triterpenes from Abies georgeiagainst QGY-7703 and LOVO tumor cell lines

	IC_{50}					
Compounds	QGY-7703	LOVO				
1	9.3	> 20.0				
2	15.2	> 20.0				
8	20.4	> 20.0				
13	14.4	> 20.0				
16	21.8	18.2				
20	7.6	14.7				
Others ^a	> 20.0	> 20.0				
Doxorubicin ^b	0.5	0.2				

in favor of 3α -hydroxylation. These assessments of the lanostane and cycloartane pharmacophore, though only preliminary, do point to clear relationships between structure and potency, and encourage the view that this unusual pharmacophore deserves further investigation.

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